STRUCTURE OF THE O-SPECIFIC POLYSACCHARIDE FROM Enterobacter cloacae STRAIN N.C.T.C. 11579 (SEROGROUP O10)

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ABSTRACT

The O-specific polysaccharide from the reference strain (N.C.T.C. 11579) for *Enterobacter cloacae* serogroup O10 has been isolated and characterised. By means of n.m.r. spectroscopy and methylation analysis, and by studies of the products obtained by Smith degradation or by *N*-deacetylation–deamination, the repeating unit of the polysaccharide could be allocated the structure shown. The polysaccharides from two cross-reacting serogroups (O9 and O11) have the same monosaccharide composition.

$$\alpha$$
-D-Glcp

1

 \downarrow

4

2) α B Monn (1 - 2) α B Monn (1 - 2) α B Evo

 \rightarrow 2)- α -D-Manp-(1 \rightarrow 2)- β -D-Manp-(1 \rightarrow 3)- α -D-FucpNAc-(1 \rightarrow 6)- α -D-Manp-(1 \rightarrow

INTRODUCTION

Enterobacter cloacae is one of a number of Gram-negative bacteria of which antibiotic-resistant strains are increasingly common¹⁻⁴ and of growing concern as opportunistic pathogens. Reliable, economical, and discriminatory methods for epidemiological studies of the species are highly desirable. Schemes of O serotypes^{5,6} and H serotypes⁵ have been drawn up, and schemes for bacteriocin^{7,8} and bacteriophage⁹ typing have also been developed. An attempt has been made¹⁰ to establish the chemical basis for serological differentiation between 14 reference strains for one O-serotyping scheme⁶ by analysis and fractionation of their lipopolysaccharides (the presumed O antigens). Three lipopolysaccharides (from strains representing the cross-reacting O9, O10, and O11 serotypes) all gave rise to polymers containing glucose, mannose, and fucosamine (2-amino-2,6-dideoxygalactose) in the same proportions. We now report the results of a structural study of the O10 polymer.

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RESULTS AND DISCUSSION

The O-specific polysaccharide (yield, 48% of the parent lipopolysaccharide) contained D-mannose (60.1%), D-glucose (18.6%), and D-fucosamine (~15%, uncorrected). The presence of an N-acetyl group (and the absence of O-acetyl substituents) were apparent from the i.r. spectrum (bands at 1645 and 1560 cm⁻¹) and the n.m.r. spectra [signals at δ 2.05 (¹H), and at δ 173.99 and 22.30 (¹³C)]. Failure of the polysaccharide to adsorb onto DEAE-Sepharose CL-6B showed the absence of an acidic sugar.

A pentasaccharide repeating-unit of mannose (3), glucose (1), and 2-acetamido-2-deoxyfucose (1) residues was clear from the n.m.r. spectra of the polymer. In the $^1\text{H-n.m.r.}$ spectrum, unresolved anomeric signals (each 1 H) at δ 5.34, 5.08, and 4.82 could be assigned to mannopyranosyl residues, and doublets (each 1 H) at δ 5.28 ($J_{1,2} \sim 4$ Hz) and 4.84 ($J_{1,2} \sim 4$ Hz) to α -pyranosyl residues of the other sugars. The spectrum also contained a methyl doublet at δ 1.24 ($J_{5,6}$ 6.5 Hz) from the 2-acetamido-2-deoxyfucose residue. Anomeric signals in the $^{13}\text{C-n.m.r.}$ spectrum (Fig. 1) at δ 102.62 ($^1J_{\text{CH}}$ 173 Hz), 101.16 ($^1J_{\text{CH}}$ 160 Hz), 100.26 ($^1J_{\text{CH}}$ 172 Hz), 99.33 ($^1J_{\text{CH}}$ 176 Hz), and 97.52 ($^1J_{\text{CH}}$ 171 Hz) showed that only one of the three mannose residues was β -linked. In support of this inference, the mannose–glucose ratio changed from 3:1 to 2:1 on oxidation of the peracetylated polymer with CrO₃. Other signals of interest in the $^{13}\text{C-n.m.r.}$ spectrum were those at δ 15.52 (C-6 of fucosamine), 48.90 (C-2 of fucosamine, diagnostic for α), and three signals for hydroxymethyl carbons at δ 60.74, 60.93, and 61.15 (showing that position 6 was substituted in one hexose residue).

The results of methylation analysis of the polysaccharide (Table I, column A) showed that the repeating unit was branched, with 2,4-disubstituted mannose at

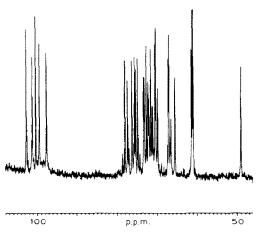


Fig. 1. 13 C-N.m.r. spectrum of the O-specific polysaccharide. In addition to the signals shown, the spectrum contained signals at δ 173.99, 22.29, and 15.52. The spectrum for the sample in D₂O was obtained at 100.61 MHz and 50°, with complete proton-decoupling and tetramethylsilane as the external reference.

the branch point. The terminal (unsubstituted) hexose was identified as glucose by h.p.l.c. of the 2,3,4,6-tetra-O-methylhexitol¹² and by g.l.c. of the 2,3,4,6-tetra-O-methylhexose 1-acetates¹³. Consistent with the results of methylation analysis, the glucose and two of the mannose residues were destroyed on periodate oxidation of the polymer, and the consumption of NaIO₄ (5.94 μ mol.mg⁻¹) was in good agreement with the value (5.99 μ mol.mg⁻¹) expected. After reduction (NaBH₄) and Smith hydrolysis of the oxidised polymer, a single product, eluted from Sephadex G-15 in about the position of a trisaccharide, was obtained. The product also gave a single spot in p.c. and (as the methylated oligosaccharide-alditol) one major peak in g.l.c.

The ¹H-n.m.r. spectrum of the Smith-degradation product contained anomeric signals (each 1 H) at δ 4.84 ($J_{1,2}$ 3.7 Hz) and 4.74 (unresolved), and methyl signals at δ 2.05 (singlet) and 1.25 ($J_{5.6}$ 6.6 Hz). The ¹³C-n.m.r. spectrum contained 17 discrete signals, including anomeric signals at δ 101.50 (${}^{1}J_{CH}$ 158 Hz) and 97.50 (${}^{1}J_{CH}$ 171 Hz) and two signals for hydroxymethyl carbons at δ 62.64 and 61.09. These data show that the periodate-resistant (branch point) mannose residue is β -linked, that it is adjacent to the 2-acetamido-2-deoxyfucose residue, and that the aglycon end of the Smith-degradation product is 1-substituted glycerol (from the original 6-substituted mannose) rather than 2-substituted glyceraldehyde (from the original 2-substituted mannose). The sequence of residues in the oligosaccharide-alditol was established by m.s. of its methylation product, giving fragment ions of m/z 219 and 187, diagnostic for terminal hexose, and of m/z 318, diagnostic for a 2-acetamido-2-deoxyfucosylglycerol residue, inter alia (Fig. 2). Confirmation was provided by the production of 1,5-di-O-acetyl-2,3,4,6-tetra-Omethylmannitol-1-d on methylation analysis of the oligosaccharide-alditol. Thus, structure 1 can be assigned to the Smith-degradation product, and structure 2 to the corresponding trisaccharide sequence in the parent polymer. An interpretation of the ¹³C-n.m.r. spectrum for the Smith-degradation product is given in Table II.

TABLE I

METHYLATION ANALYSES^a

Methylation product ^b	Relative peak area (g.l.c.)	.l.c.)
	A	В
1,4-AHdTal		+
2,3,4,6-Glc	1.03	1.50
2,3,4,6-Man		\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \
3,4,6-Man	1.00	1.00
2,3,4-Man	0.86	Trace
3,6-Man	0.76	0.85
2,4-FucNAc	+	

^aKey: A, native polymer; B, oligosaccharide-alditol from deamination reaction sequence. ^b1,4-AHdTal = 3-O-acetyl-2,5-anhydro-6-deoxy-1,4-di-O-methyltalitol; 2,3,4,6-Glc = 1,5-di-O-acetyl-2,3,4,6-tetra-O-methylglucitol, etc.

In order to locate the 2-substituted mannose residue and the lateral substituent at the branch point, the polymer was subjected to N-deacetylation and then deamination. Most of the product was eluted from Sephadex G-15 in about the position of the tetrasaccharide stachyose, depolymerisation confirming that 2acetamido-2-deoxyfucose was a component of the main chain in the polymer. Methylation analysis, monitored by g.l.c.-m.s. of the methylated alditol acetates (Table I, column B) and by g.l.c. of the methylated aldose acetates, showed that the deamination product contained non-reducing terminal residues of both glucose and mannose, the latter being derived from the residue originally 6-substituted. When the deamination product was treated with α -D-mannosidase or α -D-glucosidase, the appropriate monosaccharide was released. Methylation analysis of the α -D-glucosidase-treated material showed that the 2,4-disubstituted mannose residue had been lost, with a corresponding increase in abundance of 2-substituted mannose (but no 4-substituted mannose). From this result, it can be concluded that the original 2-substituted mannose residue was in the main chain, and that the terminal glucose residue was attached directly to the branch-point mannose residue at position 4. Thus, structure 3 can be assigned to the repeating unit of the O10 polymer.

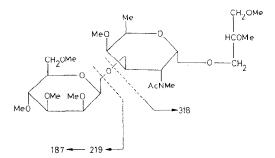


Fig. 2. Diagnostic fragment ions in the electron-impact mass spectrum of the per-O-methylated Smith-degradation product from the O-specific polysaccharide.

TABLET

Carbon atom	Chemical shift (p.p.m.)a		
¹³ C-n.m.r. data for the smi	TH-DEGRADATION PRODUCT (1)		
IABLE II			

Carbon atom	Chemical shift (p.p.m.) ^a			
	β -D-Manp-($l \rightarrow$	\rightarrow 3)- α -D-FucpNAc-(1 \rightarrow	→1)-Glycerol	
C-1	101.50	97.50	68.67 ^b	
C-2	70.55	48.45	71.30	
C-3	73.02	77.39	62.64^{b}	
C-4	66.86	70.91		
C-5	76.32	66.56	,	
C-6	61.09^{b}	15.46		
-NHC(O)CH ₃		174.49		
-NHC(O)CH ₃		22.04		

"The spectrum was recorded at 40° with tetramethylsilane as the external reference. Assignments of some signals with similar chemical shifts may be interchanged. The chemical shifts recorded were ~ 1.3 p.p.m. lower than the literature data^{14–18} used in making the assignments. ^bSignals which gave a triplet in the proton-coupled spectrum.

$$\alpha$$
-D-Glcp
1
 \downarrow
4
 \rightarrow 2)- α -D-Man p -(1 \rightarrow 2)- β -D-Man p -(1 \rightarrow 3)- α -D-Fuc p NAc-(1 \rightarrow 6)- α -D-Man p -(1 \rightarrow

The O10 antigen described above is one member of a group (O8 to O12, inclusive) of related antigens, three of which are known to be constructed from glucose, mannose, and fucosamine¹⁰. There are, presumably, structural differences between these related antigens, as monospecific O antisera can be prepared by absorption with appropriate boiled cells⁶. The presence of an *O*-acyl substituent in the O11 antigen may contribute to the specificity of the antigen¹⁰.

EXPERIMENTAL

Growth of bacteria, and isolation and fractionation of the lipopolysaccharide. — E. cloacae N.C.T.C. 11579 (the reference strain for the O10 serogroup⁶) was grown for 16 h at 37° in Nutrient Broth No. 2 (Oxoid) with aeration. From 123 g of wet cells, 3.6 g of cell walls were obtained, as in related studies¹⁹, and used for the extraction of lipopolysaccharide (1.4 g). The water-soluble products of mild acid hydrolysis (1% acetic acid, 1.5 h, 100°) were fractionated on Sephadex G-50 and DEAE-Sepharose CL-6B (ref. 19).

General methods. — I.r. spectra were recorded with a Unicam SP-200

spectrophotometer for samples dispersed in KBr. Optical rotation was determined with a Bendix polarimeter (Model 143A). N.m.r. spectra were recorded for solutions in D₂O with Bruker WH-400 spectrometers. ¹³C-N.m.r. spectra (with complete proton-decoupling, with gated decoupling, or with the INEPT pulse sequence) were recorded at 40° or 50° with tetramethylsilane as the external reference. ¹H-N.m.r. spectra were recorded at 80° or 85° with sodium 3-trimethylsilylpropanoate-d₄ as the internal reference. G.l.c. was carried out either with a packed column of OV-275 in a Pye Unicam 104 chromatograph, or with fused-silica capillary columns of BP1 or BP10 in a Carlo Erba Mega 5160 chromatograph. Reverse-phase h.p.l.c. of 2,3,4,6-tetra-O-methylhexitols was carried out with aqueous acetonitrile¹². G.l.c.-m.s. was carried out with a Finnigan model 1020B instrument, either in the e.i. mode or by c.i. with methane as the reagent gas²⁰. The solvent systems and detection reagents used in p.c. were those described previously¹⁹. High-voltage paper electrophoresis was done at pH 5.3 with the buffer system pyridine-acetic acid-water (5:2:43).

Determination of monosaccharide composition. — For neutral sugars, samples were hydrolysed with 2M HCl for 2 h at 105°. After neutralisation of the acid and deionisation, hexoses in the hydrolysates were identified and determined by p.c., by enzymic assay¹⁹, and by g.l.c. of the alditol acetates.

For amino sugars, samples were hydrolysed with 4M HCl for 6 h at 105° under nitrogen. Fucosamine was identified by p.c. of the compound and its N-acetyl derivative, by high-voltage electrophoresis and autoanalysis, by a positive Elson-Morgan reaction²¹, and by g.l.c.-m.s. of the derived 2,5-anhydro-6-deoxytalitol triacetate¹⁰. It was assigned the D configuration from the positive optical rotation of the hydrochloride.

Degradative methods. — Standard procedures were used for the preparation²², purification²³, and hydrolysis²⁴ of methylated poly- and oligo-saccharides. The products were analysed as methylated alditol acetates (by g.l.c.-m.s.), methylated aldose acetates (by g.l.c.¹³), methylated alditols (by h.p.l.c.¹²), or the methylated oligosaccharide-alditol (by m.s.). Periodate oxidation of the O-specific polysaccharide (28 mg) was carried out with 50mm NaIO₄ for 5 days at 4°. The consumption of NaIO₄ was monitored by the method of Avigad²⁵. After a conventional work-up and reduction (NaBH₄), the oxidised polymer was hydrolysed¹⁹ and the Smith-degradation product isolated by chromatography on Sephadex G-15. The N-deacetylation of reduced (NaBH₄) polysaccharide (40 mg) with alkali in aqueous Me₂SO²⁶ was carried out for 16 h at 95°, and the purified product was deaminated²⁷, then reduced (NaBH₄), and chromatographed on Sephadex G-15. Enzymic hydrolyses¹⁹ of the deamination product were carried out with α -Dmannosidase (EC 3.2.1.24, from jack bean, Sigma) or α -D-glucosidase (EC 3.2.1.20, from rice, Sigma), and were monitored by p.c. and by enzymic assay of the monosaccharides released. Peracetylation and CrO₃ oxidation (1 h, 50°) of the O-specific polysaccharide were done by standard methods²⁷. O-Acetyl groups were removed by treatment with methanolic 0.2M KOH for 2.5 h at room temperature.

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